

# Experimental speed of sound in CO<sub>2</sub>-rich mixtures with methanol. Extrapolation to pure CO<sub>2</sub>

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## High pressure speed of sound apparatus

Double-path double-echo pulsed ultrasonic system

253 K – 473 K  $u(T)=0.015$  K

0.1 MPa – 200 MPa  $u(P)=0.02$  MPa

**Frequency: 5 MHz**

## At 5 MHz

Pure CO<sub>2</sub> → no signal

CO<sub>2</sub>-rich mixtures of interest for CCS, → poor or no signal

C. W. Lin, PhD Thesis, 2013

Pure CO<sub>2</sub>: No signal at 2 MHz

Partial success at 0.5 MHz

C.W. Lin, PhD Thesis, 2013

C.W. Lin and J.P.M. Trusler, 2014, J. Chem. Eng. Data, 59, 4099-4109

**Doping CO<sub>2</sub> with small amounts of propane at 2 MHz**

Good signals

Successful extrapolation to pure CO<sub>2</sub>

## At 5 MHz

CO<sub>2</sub> + propane is opaque for  $x_{\text{CO}_2} > 0.8$  (approx.)

# In our study CO<sub>2</sub>+SO<sub>2</sub>

**We hoped**

**SO<sub>2</sub> itself could act as doping agent in the CO<sub>2</sub>+SO<sub>2</sub> mixtures**

**BUT**

**It only works at  $x_{\text{CO}_2} \leq 0.9$**

**At  $x_{\text{CO}_2} = 0.95$ :**

**very poor signals along very short ranges of pressure and only at low temperatures**

# LOOKING FOR A SUITABLE DOPING AGENT

## Methanol

- reduces significantly the sound absorption coefficient of the mixture
- works well until  $x_{\text{CO}_2} \approx 0.99$  in a  $\text{CO}_2 + \text{CH}_3\text{OH}$  mixture
- can appear in the CCS facilities
  - ✓ impurity in anthropogenic  $\text{CO}_2$
  - ✓ used to avoid hydrates formation
  - ✓ a residue from pipeline drying

**Experimental speed of sound in CO<sub>2</sub>-rich mixtures with methanol.  
Extrapolation to pure CO<sub>2</sub>**

## **Aim of this work**

**To test the suitability of methanol as doping agent, in order to obtain good measurements of speed of sound in CO<sub>2</sub> at 5 MHz and, in the future, in CO<sub>2</sub>-rich mixtures of interest for CCS.**

**To evaluate the doping effect.**

Experimental speed of sound in CO<sub>2</sub>-rich mixtures with methanol.  
Extrapolation to pure CO<sub>2</sub>

## For this purpose

- We measured the **speed of sound** in seven CO<sub>2</sub>-rich mixtures with methanol at several  $P$  and  $T$  and at **5 MHz**
- We obtained **extrapolated** values of  $c$  in **pure CO<sub>2</sub>**
- We evaluated the **effect of the doping** on  $c$  by comparing our results with the Span and Wagner EoS
- We used the experimental results for the mixtures to **validate** the PC-SAFT and the GERG EoSs

# Measuring of the speed of sound, $c$ , in seven mixtures $\text{CO}_2 + \text{CH}_3\text{OH}$

COMPOSITIONS
$x_{\text{CO}_2}$
0.8005
0.9025
0.9503
0.9700*
0.9794
0.9845
0.9898

TEMPERATURES (K)
263.15
298.15
323.15

PRESSURES (MPa)
3-200

$u(c) = 0.059\%$

Used for the extrapolation to pure  $\text{CO}_2$

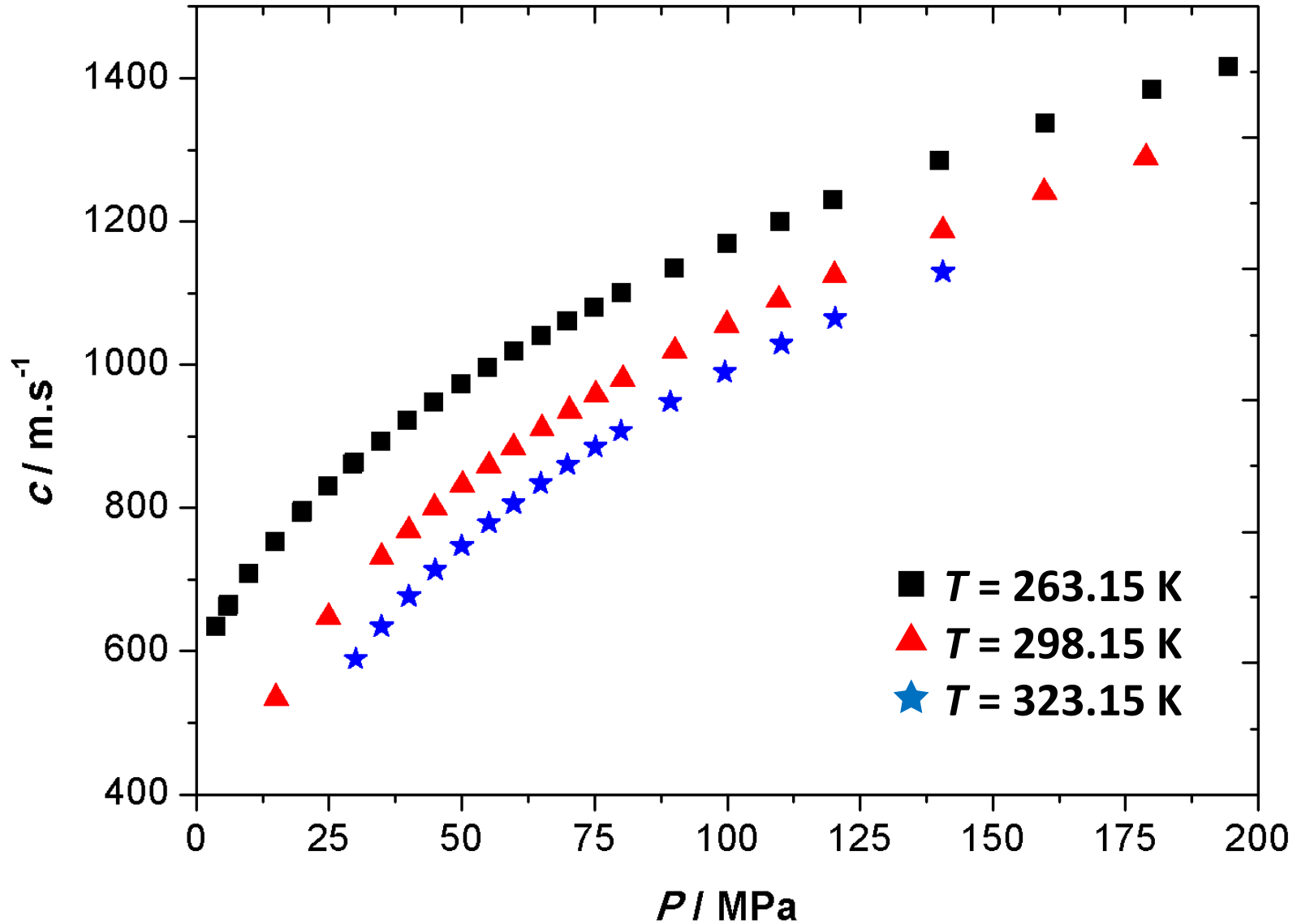
Compared with the Span and Wagner EoS

All the isotherms were used for PC-SAFT and GERG EoS validation

\*At 263.15 K, Rivas, C., Gimeno, B., Bravo, R., Artal, M., Fernández, J., Blanco, S.T., Velasco, I., 2nd International Forum on Recent Developments of CCS Implementation, Athens 2015.



# Experimental speeds of sound, $c$ , in the $\text{CO}_2 + \text{CH}_3\text{OH}$ mixture with $x_{\text{CO}_2} = 0.9503$ .



# With the experimental results of the five most concentrated mixtures

COMPOSITIONS
$x_{\text{CO}_2}$
0.8005
0.9025
0.9503
0.9700
0.9794
0.9845
0.9898

TEMPERATURES (K)
263.15
298.15
323.15

PRESSURES (MPa)
3-200

**EXTRAPOLATION TO  
PURE CO<sub>2</sub>**

The experimental results of each isotherm and isopleth were fitted to a polynomial:

$$(P - P^\#) = a_1(c - c^\#)^1 + a_2(c - c^\#)^2 + a_3(c - c^\#)^3 \quad (1)$$

$$P^\# = 70 \text{ MPa}; c^\# = c \text{ at } P^\#$$

Coefficients of the equation 1 for the mixture CO<sub>2</sub> + CH<sub>3</sub>OH with  $x_{\text{CO}_2} = 0.9503$  at temperatures  $T$ , and mean relative deviations.

$x_{\text{CO}_2}$	$T/\text{K}$	$10 \times a_1$ MPa.m <sup>-1</sup> .s	$10^4 \times a_2$ MPa.m <sup>-2</sup> .s <sup>2</sup>	$10^8 \times a_3$ MPa.m <sup>-3</sup> .s <sup>3</sup>	$MRD_c$ %
0.9503	263.15	2.5039	2.543	7.33	0.019
	298.15	2.1649	2.271	7.40	0.012
	323.15	2.0001	2.127	7.10	0.004

$$MRD_c(\%) = \frac{100}{N} \sum_{i=1}^N \left| \frac{c_{i,fit} - c_i}{c_i} \right|$$

From the polynomials (1), we obtained values of  $c$  at round values of  $P$  at each  $T$  and at each composition

These values of  $c$ , at each  $T$  and at each round  $P$ , for the five studied compositions, were fitted to

$$c(x_{\text{CO}_2}) = c_0 + c_1(1 - x_{\text{CO}_2}) + c_2(1 - x_{\text{CO}_2})^2 \quad (2)$$

$c_0$  is the speed of sound at a given  $T$  and at a given round value of  $P$  for  $x_{\text{CO}_2} = 1$

$c_0$  is the extrapolated speed of sound in pure  $\text{CO}_2$  at  $T$  and  $P$

Average overall uncertainty 0.12%

That way, we obtained the extrapolated speeds of sound in pure  $\text{CO}_2$  at each given  $T$  and round  $P$

# Derived $c$ in pure $\text{CO}_2$ : comparison with the literature and the S-W EoS

Reference	Type of data	$MRD_c(\%)$
Pitevskaya and Bilevich, 1973	Direct experimental	0.61%*
Al-Siyabi, 2013	Direct experimental	Consistent**
Lin, 2014	Direct experimental at 0.5 MHz	0.44%*
Lin and Trusler, 2014	Derived with propane at 2 MHz	0.21%*
<b>Span and Wagner, 1996</b>	<b>Equation of state</b>	<b>0.43%*</b>
<b>Tolerance margin of the S-W EoS</b>		<b>0.5%-2%</b>

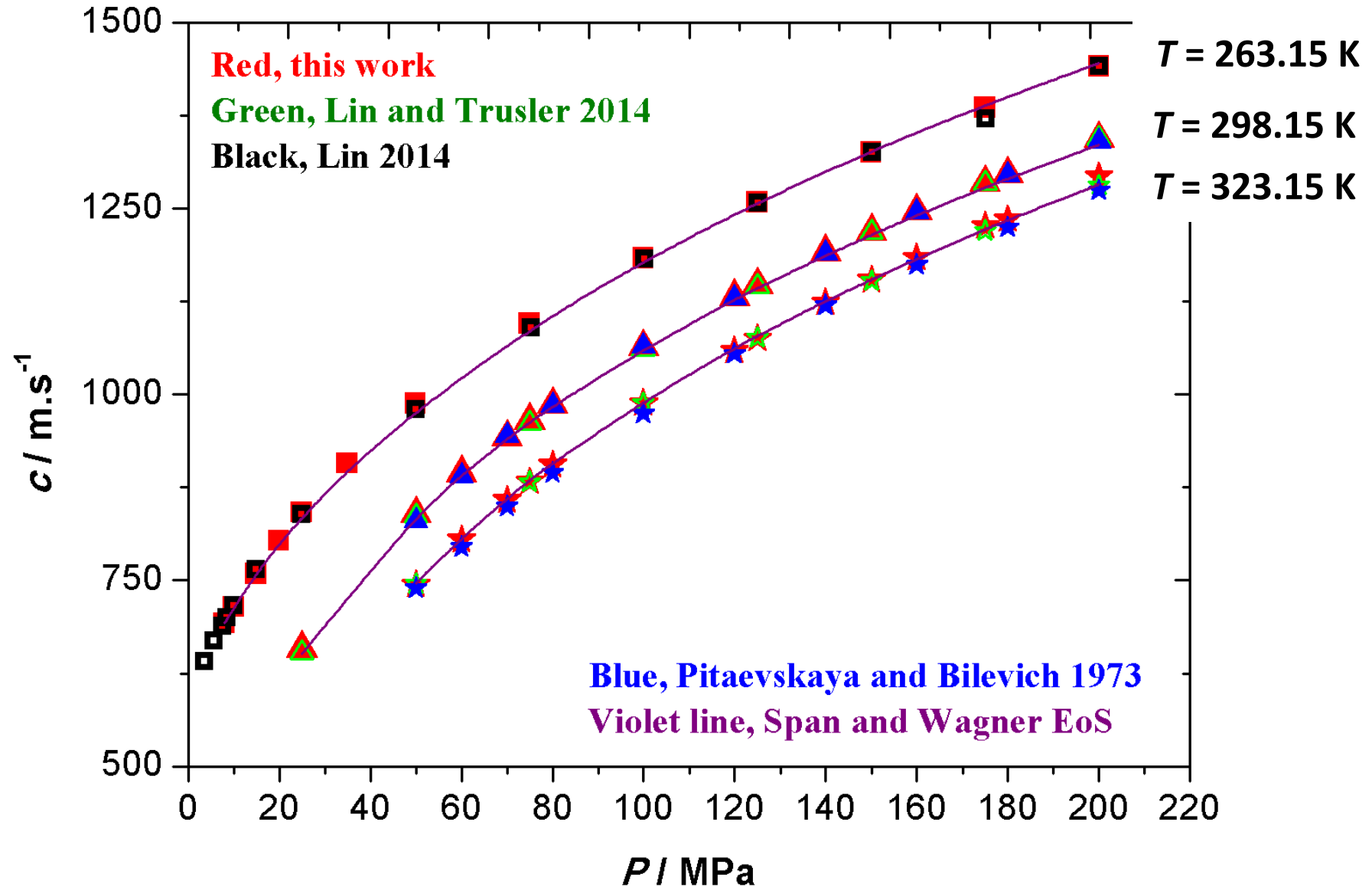
\*At the common temperatures

\*\*Temperatures were different

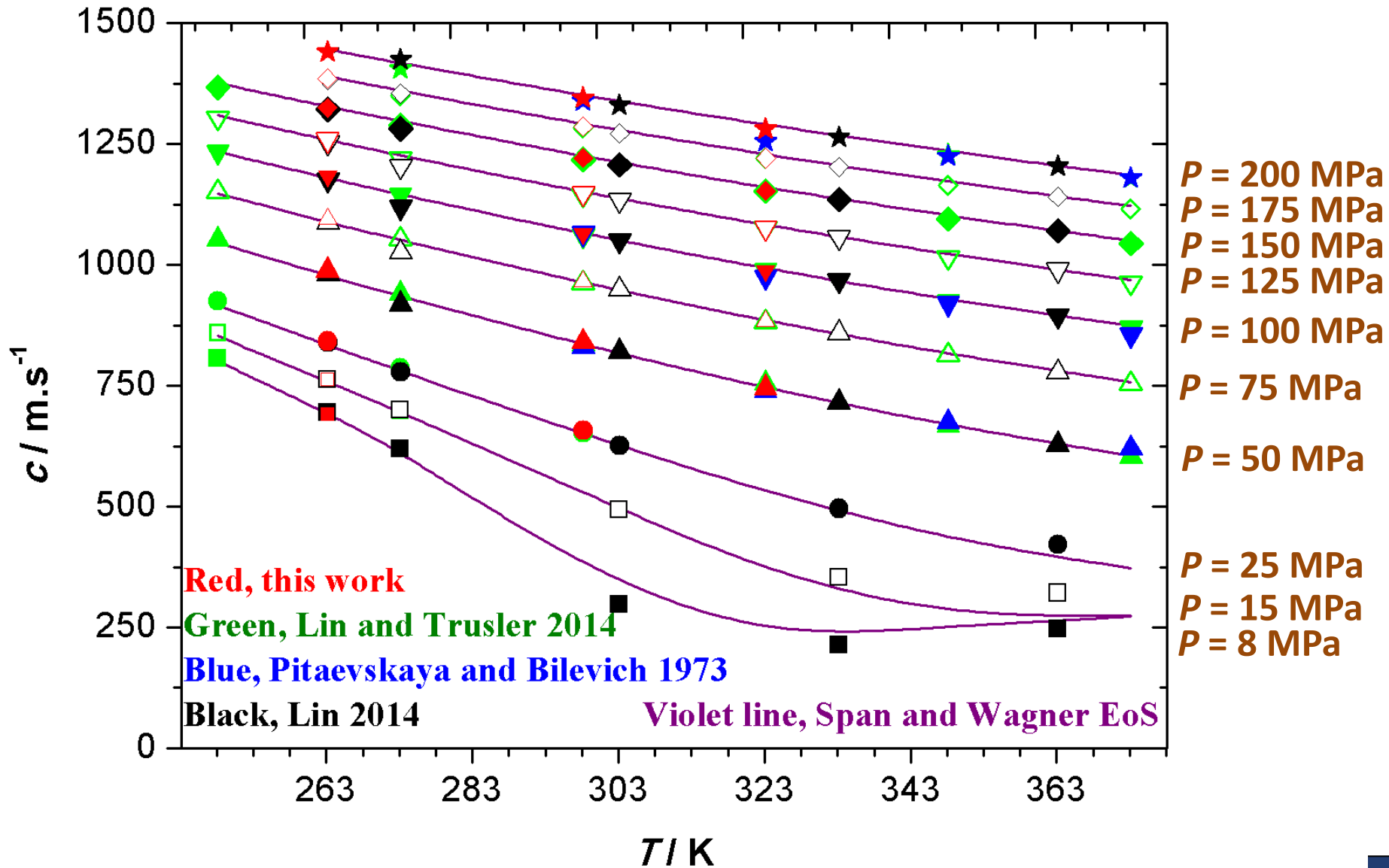
$$MRD_c(\%) = \frac{100}{N} \sum_{i=1}^N \left| \frac{c_{i,lit} - c_i}{c_i} \right|$$

**The deviations are higher than the uncertainty of our  $c$  in pure  $\text{CO}_2$ , (0.12%), but lower than the tolerance margin of the Span and Wagner EoS**

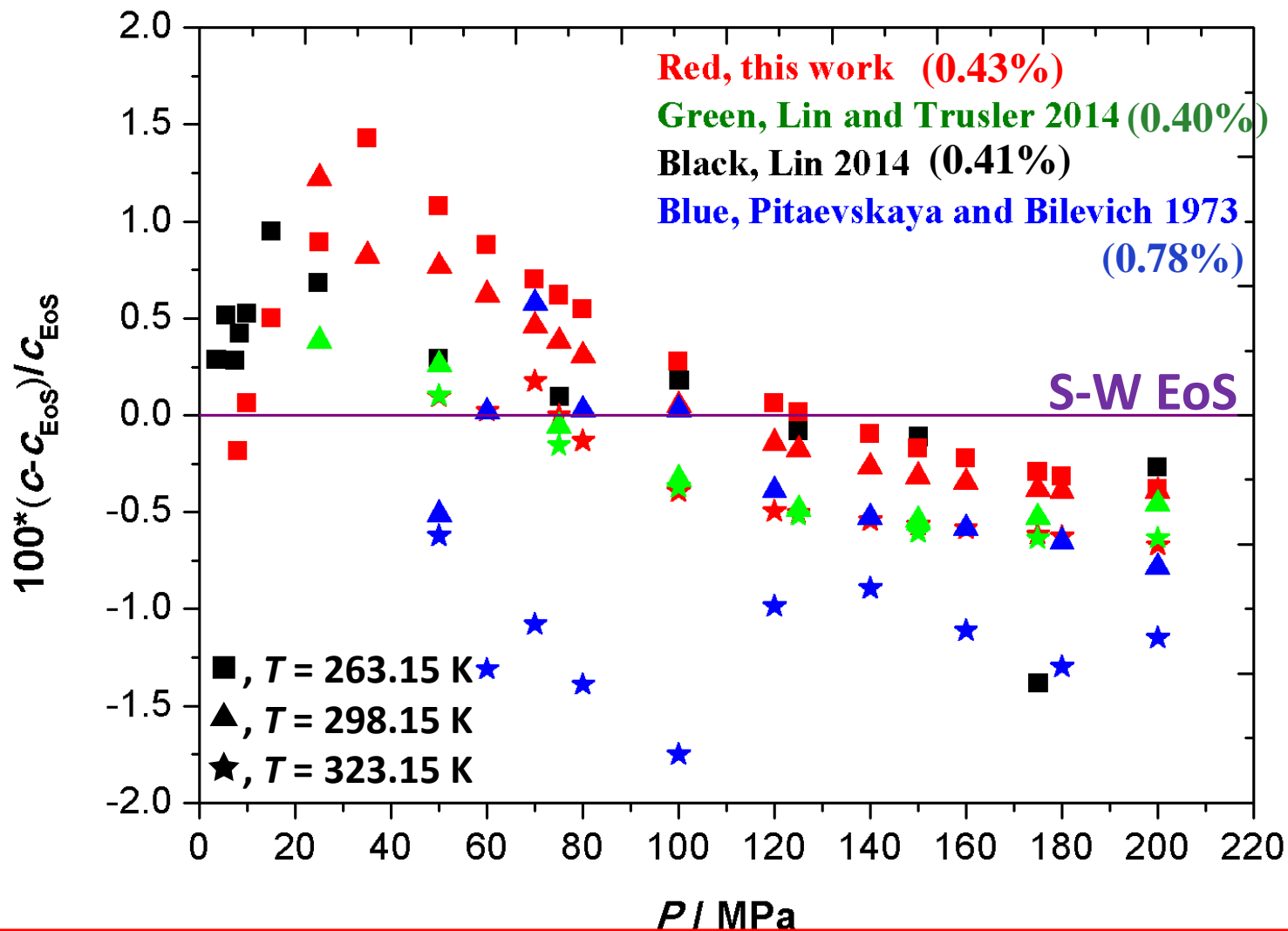
# Speed of sound, $c$ , in pure CO<sub>2</sub> versus pressure, $P$ .



# Speed of sound, $c$ , in pure CO<sub>2</sub> versus temperature, $T$ .



# Relative deviations of $c$ in pure $\text{CO}_2$ from the Span and Wagner EoS

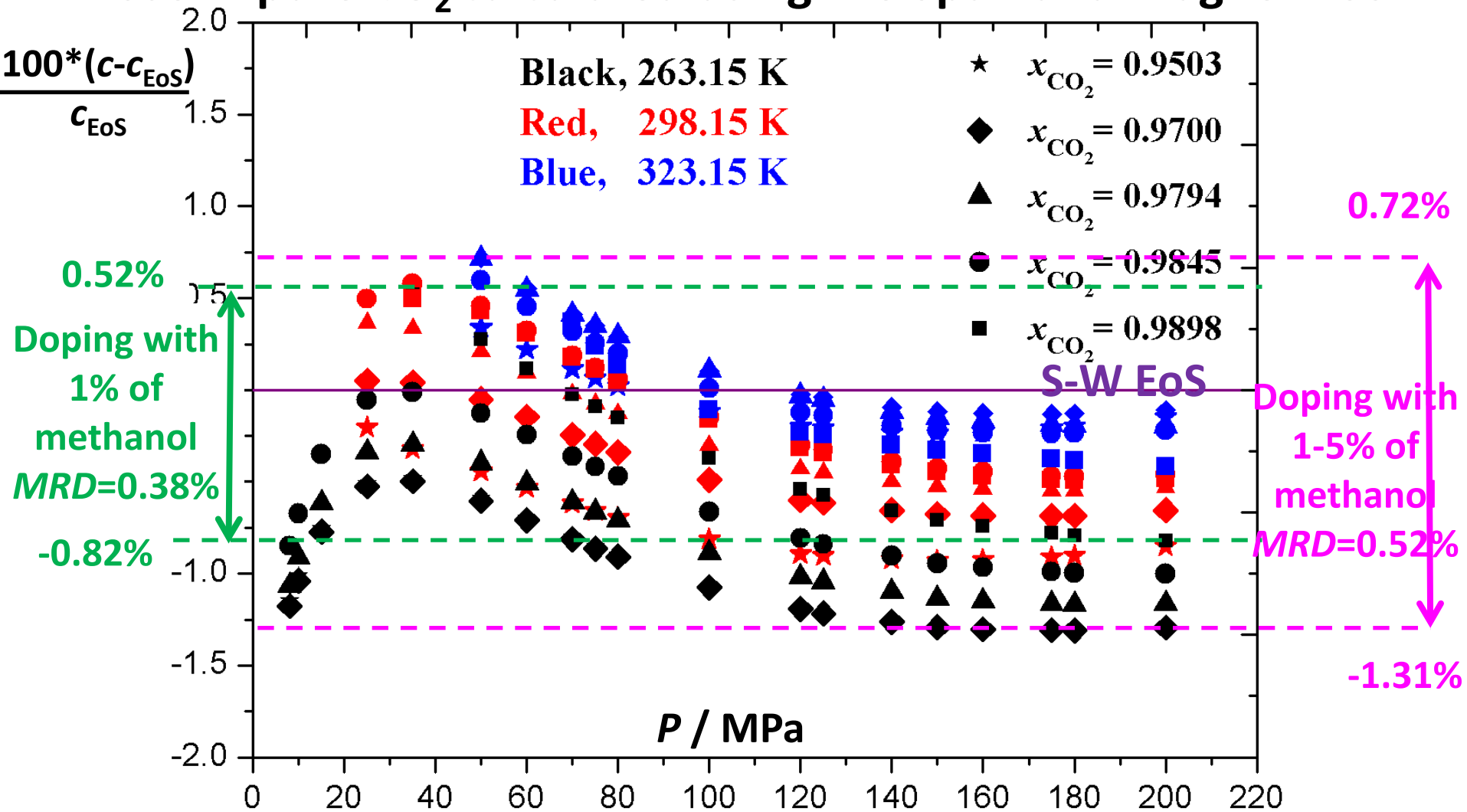


**LAST THREE FIGURES:**

**Good agreement of the whole set of data and the Span and Wagner EoS**



# Relative deviations of our $c$ in the mixtures $\text{CO}_2+\text{CH}_3\text{OH}$ from those in pure $\text{CO}_2$ calculated using the Span and Wagner EoS.



The deviations are higher than the experimental uncertainty (0.059%), but they are almost always inside the tolerance margin of the Span and Wagner EoS (0.5-2%)

# COMPARISON WITH EoSs

OUR EXPERIMENTAL SPEEDS OF SOUND FOR  
ALL THE CO<sub>2</sub>+CH<sub>3</sub>OH MIXTURES

-

PC-SAFT AND GERG EoSs

# PC-SAFT EoS applied to CO<sub>2</sub>+CH<sub>3</sub>OH

$$\tilde{a} = \tilde{a}^{id} + \tilde{a}^{hc} + \tilde{a}^{dis} + \tilde{a}^{assoc} + (\tilde{a}^{QQ} + \tilde{a}^{DD} + \tilde{a}^{QD})$$

- For pure compounds  $m$ ;  $\sigma$ ; and  $\varepsilon$  calculated from the pure compounds' critical temperatures and pressures (Gil et al., 2012)\*.
- The mixing parameters  $\sigma_{ij}$  and  $\varepsilon_{ij}$  :

$$\sigma_{ij} = \frac{1}{2} (\sigma_i + \sigma_j) \quad \varepsilon_{ij} = \sqrt{\varepsilon_i \varepsilon_j} (1 - k_{ij})$$

$$k_{ij} = -0.323 + 2.88 \times 10^{-4} T$$

(\*) Gil, L. et al., 2012. The Journal of Supercritical Fluids, 71, 26-44.

# PC-SAFT EoS applied to CO<sub>2</sub>+methanol

non-self-association  
CO<sub>2</sub>



self-association compound  
Methanol

## INDUCED ASSOCIATION\*:

- The association volume,  $\kappa^{A_i B_i} = \kappa^{methanol}$ , and the association energy,  $\varepsilon^{A_i B_i} = 0$  for **CO<sub>2</sub>** with a **2C** association scheme.
- The association volume,  $\kappa^{A_i B_i}$ , and the association energy,  $\varepsilon^{A_i B_i}$  for **methanol** with a **2B** association scheme.
- The cross-association parameters,  $\kappa^{A_i B_j} = \kappa^{methanol}$ , and  $\varepsilon^{A_i B_j} = \kappa^{methanol} / 2$ .

(\*) Kleiner, M., Sadowski, G., 2007. The Journal of Physical Chemistry C, 111, 15544-15553.

# GERG EoS applied to CO<sub>2</sub>+CH<sub>3</sub>OH

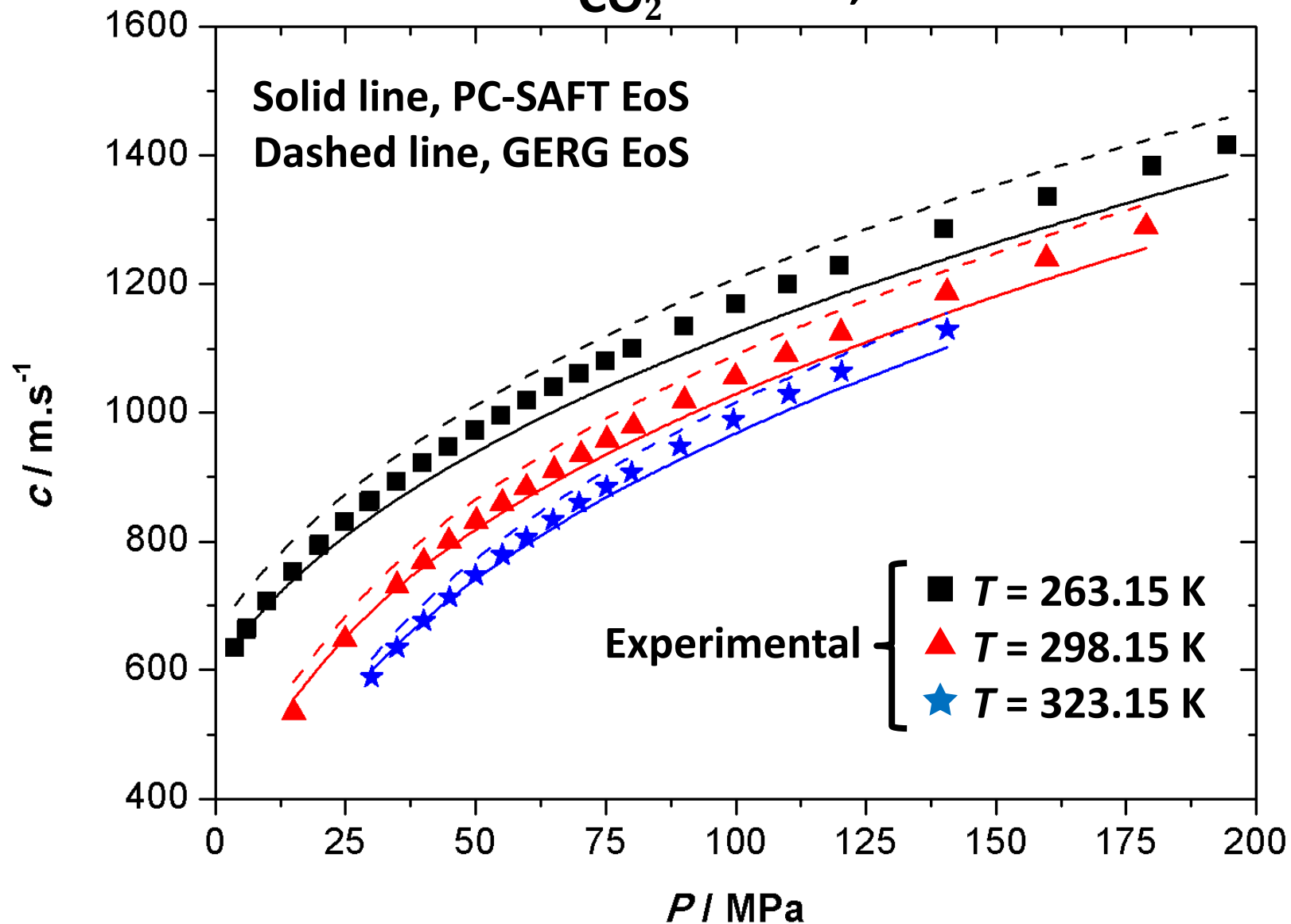
$$\tilde{a} = \tilde{a}^{id} + \tilde{a}^{res} = \sum_{i=1}^N x_i [\tilde{a}_i^{id} + \ln x_i] + \sum_{i=1}^N x_i \tilde{a}_i^{res} + \Delta \tilde{a}^{res}$$

- Although methanol is not one of the 21 compounds included in the Kunz and Wagner article (Kunz and Wagner, 2012), it has been implemented in the used **REFPROP 9** software.

(\*) Kunz, O., Wagner, W., 2012. The Journal of Chemical & Engineering Data, 57, 3032-3091.

# Speeds of sound, $c$ , in the $\text{CO}_2+\text{CH}_3\text{OH}$ mixture with

$$x_{\text{CO}_2} = 0.9503,$$



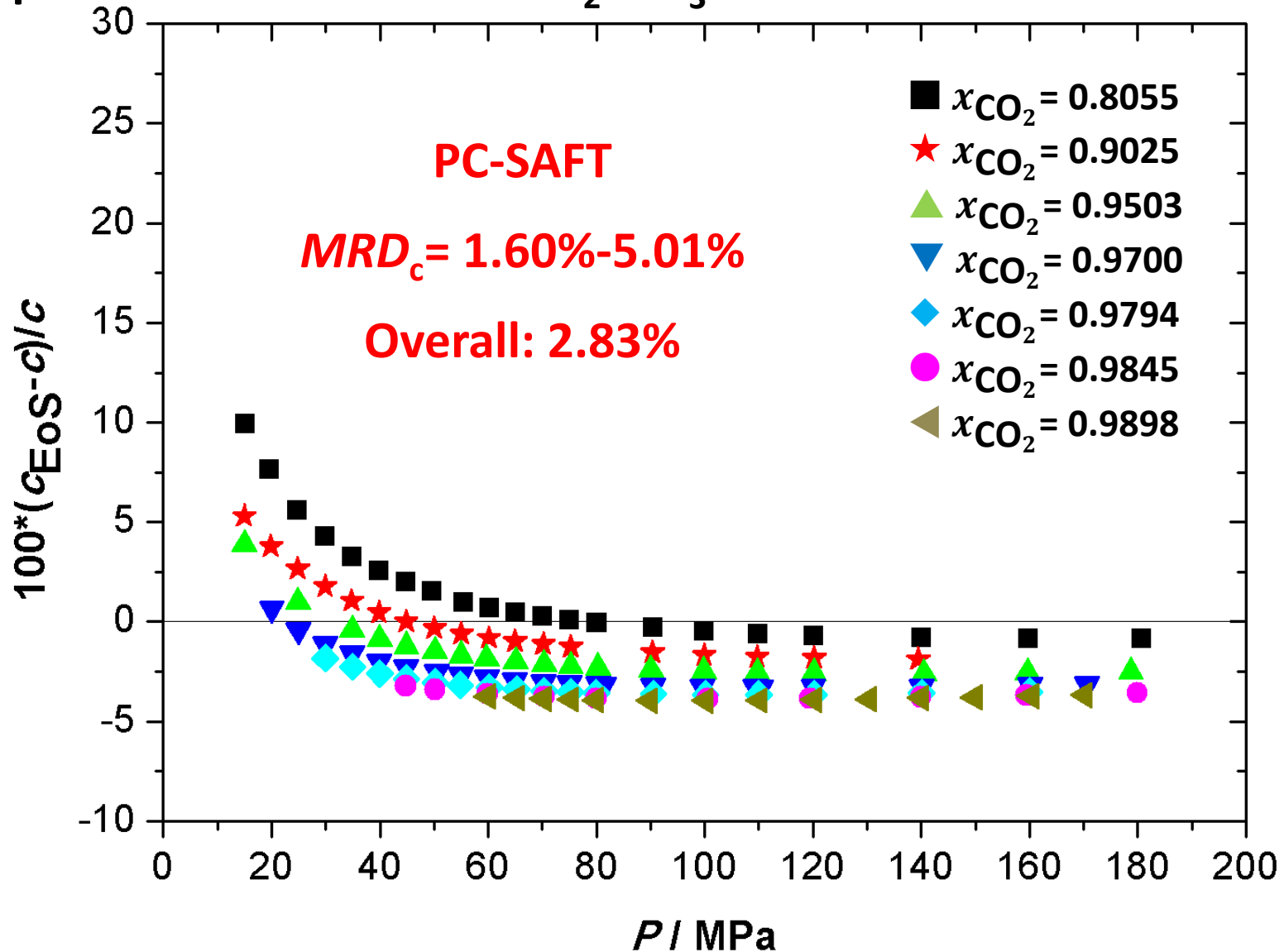
		$MRD_c(\%)$ $T = 263.15 \text{ K}$		$MRD_c(\%)$ $T = 298.15 \text{ K}$		$MRD_c(\%)$ $T = 323.15 \text{ K}$	
$x_{\text{CO}_2}$		PC-SAFT	GERG	PC-SAFT	GERG	PC-SAFT	GERG
<b>0.8005</b>		2.12	11.7	2.09	12.9	2.16	11.0
<b>0.9025</b>		2.56	7.12	1.60	7.65	1.57	6.49
<b>0.9503</b>		2.82	4.95	2.05	4.23	1.55	3.16
<b>0.9700</b>		3.08	3.62	2.61	2.29	2.56	1.40
<b>0.9794</b>		3.98	2.15	3.22	1.34	2.95	0.86
<b>0.9845</b>		4.32	1.55	3.66	0.85	3.09	0.74
<b>0.9898</b>		5.01	0.80	3.86	0.64	3.15	0.65

	$MRD_c(\%)$	
	PC-SAFT	GERG
<b>overall</b>	<b>2.83</b>	<b>4.97</b>
<b><math>x_{\text{CO}_2} \geq 0.98</math></b>	<b>3.69</b>	<b>1.06</b>

$$MRD_c(\%) = \frac{100}{N} \sum_{i=1}^N \left| \frac{c_{i,\text{EoS}} - c_i}{c_i} \right|$$

# COMPARISON EXPERIMENTAL - PC-SAFT

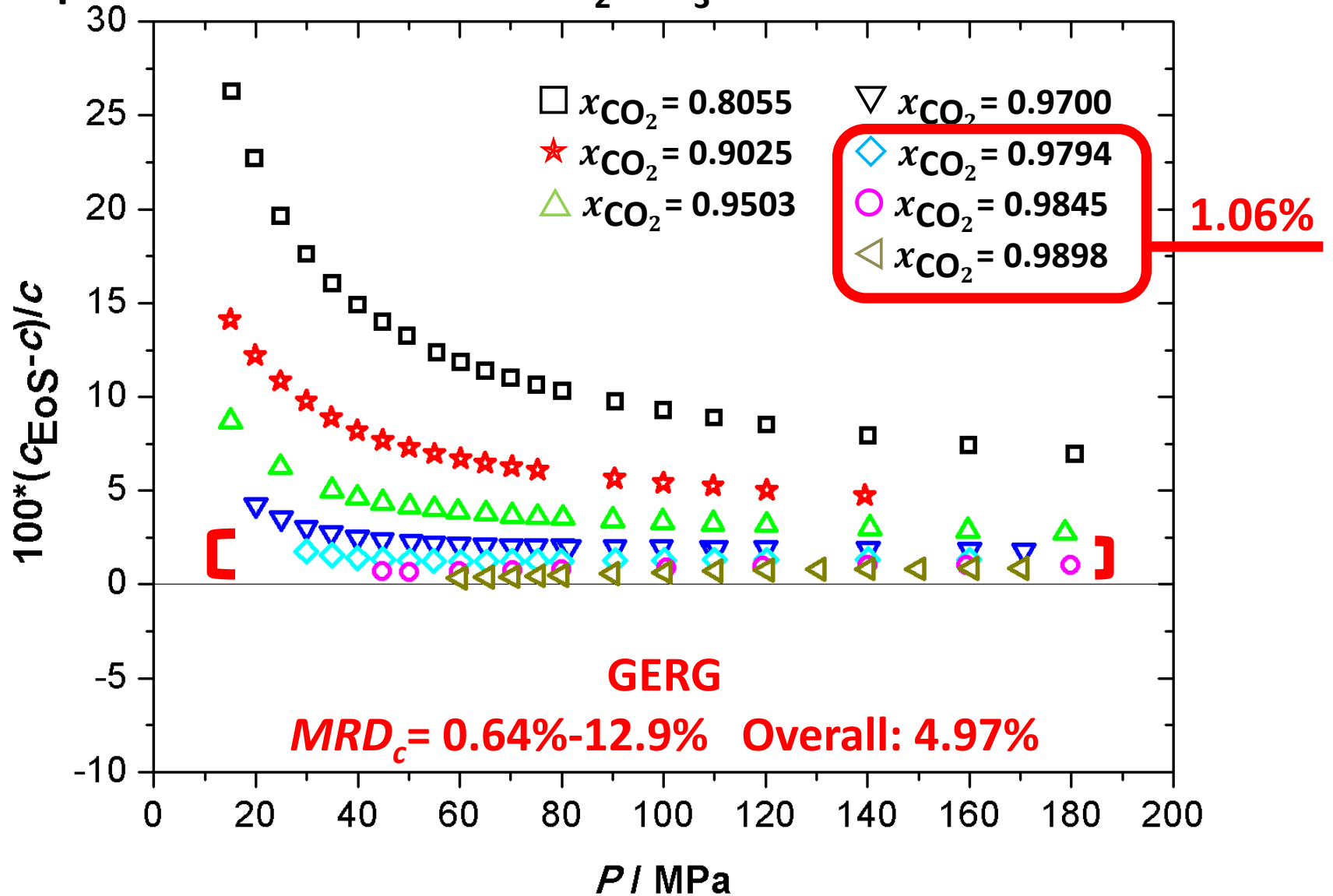
Speeds of sound for the  $\text{CO}_2 + \text{CH}_3\text{OH}$  mixtures at  $T = 298.15 \text{ K}$





# COMPARISON EXPERIMENTAL - GERG

Speeds of sound for the CO<sub>2</sub>+CH<sub>3</sub>OH mixtures at  $T = 298.15$  K



# CONCLUSIONS

Acoustic measurements are needed for CCS

Great difficulties to determine the speed of sound in pure CO<sub>2</sub> and in CO<sub>2</sub>-rich mixtures at high pressures.

Doping with methanol  $\Rightarrow$  good measures of  $c$  in CO<sub>2</sub> at **5 MHz**

The effect of the doping has been quantified.

The PC-SAFT and GERG EoS have been validated with the exception of GERG for the methanol-richest mixtures.

## NEXT STEP

To extend this method to opaque CO<sub>2</sub>-rich mixtures of interest for CCS technology.

# Speeds of sound in CO<sub>2</sub> + SO<sub>2</sub> mixtures doping with **0.8% of methanol**

$$x_{\text{SO}_2} = 0.103$$

Without methanol  
**undoped**

with methanol  
**doped**

$$\begin{aligned} x_{\text{CO}_2} &= 0.8969 \\ x_{\text{SO}_2} &= 0.1031 \end{aligned}$$

$$u_c = 0.06\%$$

$$\left. \begin{aligned} x_{\text{CO}_2} &= 0.8888 \\ x_{\text{CH}_3\text{OH}} &= 0.0080 \\ x_{\text{SO}_2} &= 0.1032 \end{aligned} \right\} x_{\text{solvent}} = 0.8968$$

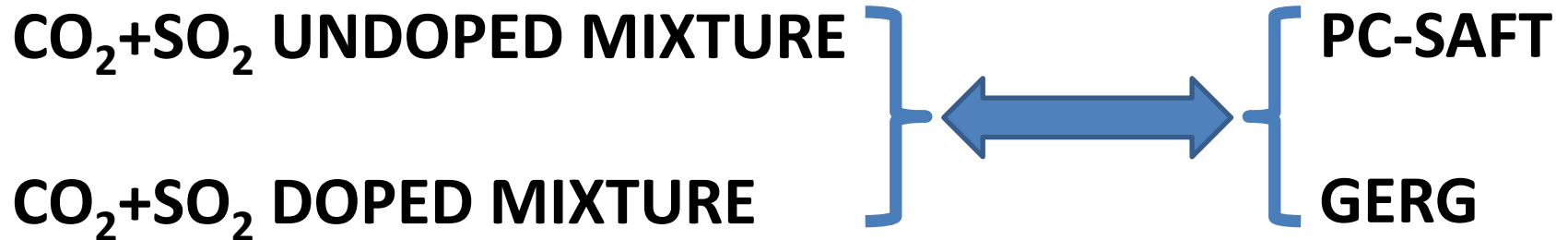
$$u_c = 0.09\%$$

Differences ( $MRD_c$ )  
**doped-undoped**

0.20% at 263.15 K  
0.14% at 293.15 K  
0.15% at 353.15 K

$$\overline{MRD_c} = 0.16\%$$

# COMPARISON



**GERG: REFPROP 9**

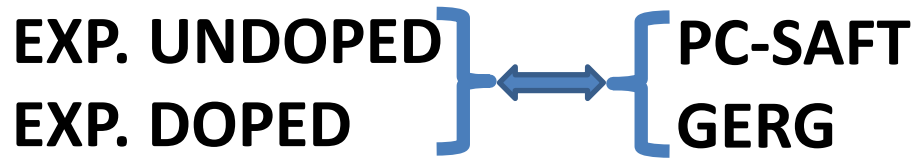
## PC-SAFT: Binary mixture

Parameter	Reference
CO <sub>2</sub>	Gross-Sadowski, 2001
SO <sub>2</sub>	Gross-Sadowski, 2001
$k_{ij}(\text{CO}_2\text{-SO}_2) = 0.03$	Diamantonis, 2013

## PCSAFT: Ternary mixture

Parameter	Reference
CO <sub>2</sub> (ASSO.2C)	Gil et al., 2012
SO <sub>2</sub>	Gross-Sadowski, 2001
CH <sub>3</sub> OH (ASSO.2B)	Gil et al., 2012
$k_{ij}(\text{CO}_2\text{-SO}_2) = 0.03$ $k_{ij}(\text{CO}_2 - \text{CH}_3\text{OH}) = -0.323 + 2.88 \times 10^{-4} T$ $k_{ij}(\text{SO}_2 - \text{CH}_3\text{OH}) = 0$	Diamantonis, 2013 Gil et al., 2012

COMPARISON



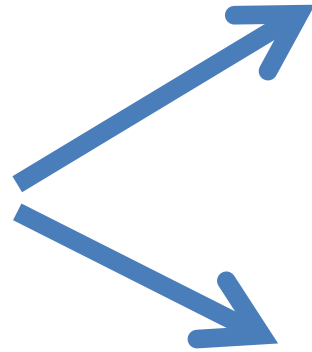
(T = 263.15 K,  
 293.15 K,  
 353.15 K)

<b><math>MRD_c(\%)</math></b>	EoS →	<u>undoped mixture</u> CO <sub>2</sub> (0.8969) + SO <sub>2</sub> (0.1031)		<u>doped mixture</u> CO <sub>2</sub> (0.8888) + SO <sub>2</sub> (0.1032) + CH <sub>3</sub> OH(0.080)	
EXPERIMENTAL ↓					
<u>undoped mixture</u> CO <sub>2</sub> (0.8969) + SO <sub>2</sub> (0.1031)		PC-SAFT GERG	2.20% 1.77%	PC-SAFT GERG	4.30% 1.58%
<u>doped mixture</u> CO <sub>2</sub> (0.8888) + SO <sub>2</sub> (0.1032)+ CH <sub>3</sub> OH(0.080)		PC-SAFT GERG	2.44% 1.23%	PC-SAFT GERG	3.72% 1.26%

$$MRD_c(\%) = \frac{100}{N} \sum_{i=1}^N \left| \frac{c_{i,EoS} - c_i}{c_i} \right|$$

# CONCLUSION

Differences  
doped - undoped  
(0.16%)



little higher than  
experimental uncertainty  
(0.09%)

much lower than  
deviations from the  
PC-SAFT and GERG EoSs  
(1.23-4.40%)

The effect of methanol in the speed of sound values is quite small for the experiments and negligible for modelling

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# Universidad Zaragoza

Título de la presentación

Subtítulo

Fecha